



SPACE PHYSICS

Lecture 3

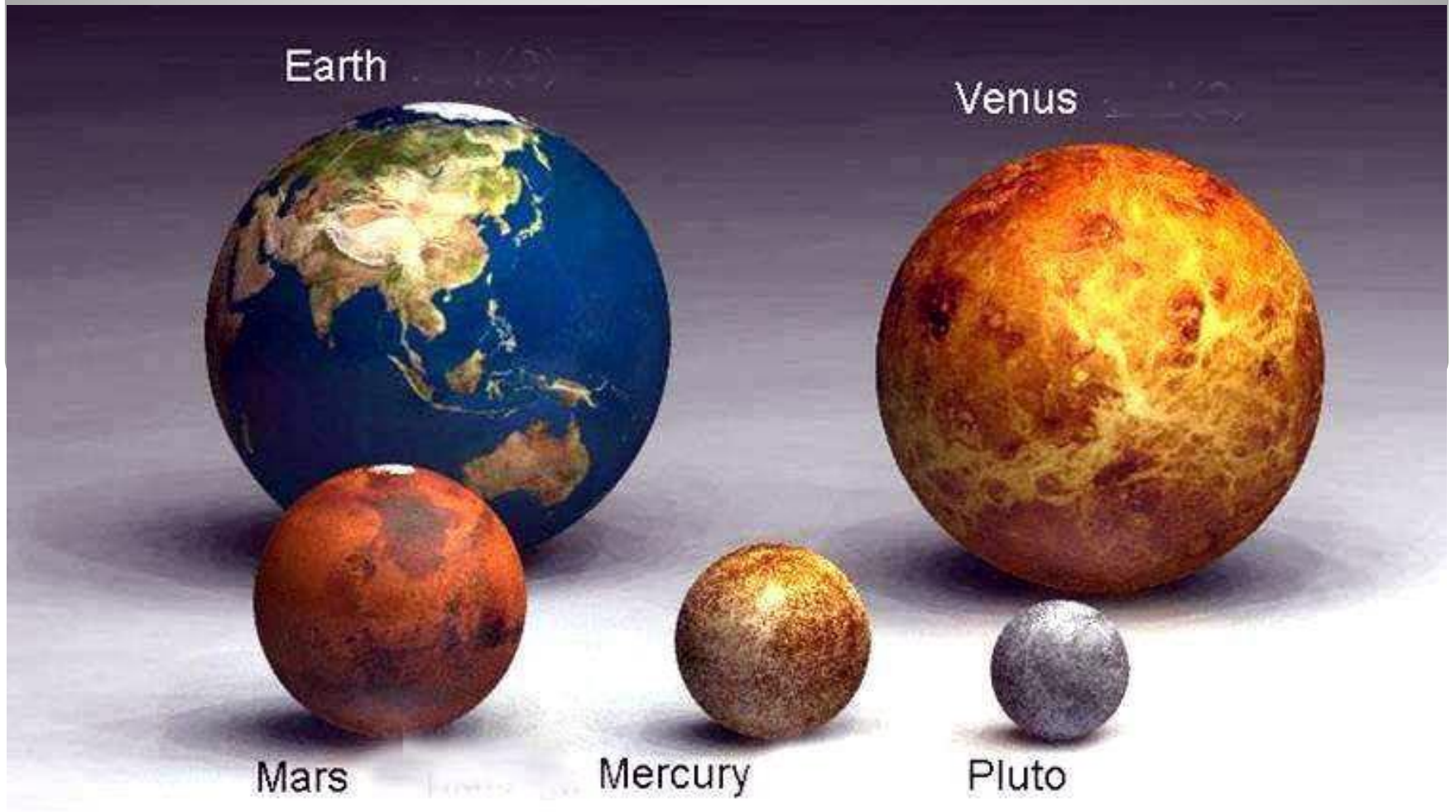


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Sun and Planets in the Solar System



Jupiter

Saturn

Uranus

Neptune

Earth

Pluto



Sun

Jupiter

Earth

Pluto





Sun



Sirius



Pollux



Arcturus

Jupiter is about 1 pixel in size

Earth is invisible at this scale

Betelgeuse

Antares

Sun (1 pixel)
Jupiter is invisible at this scale
Sirius Pollux Arcturus



Rigel

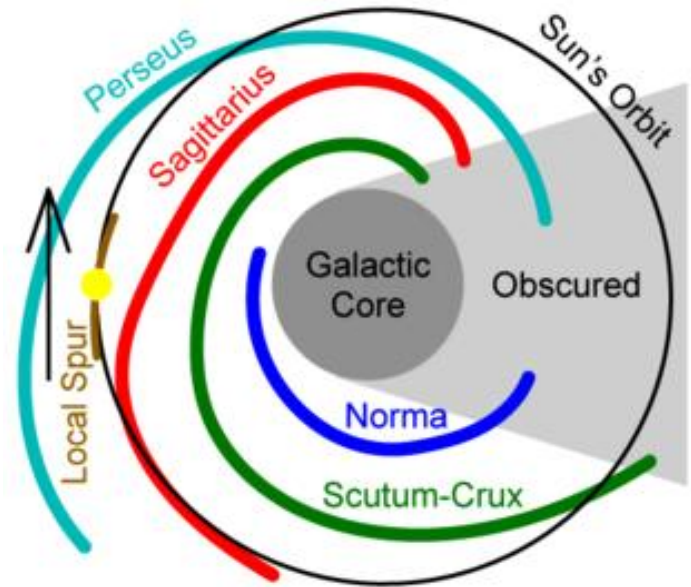
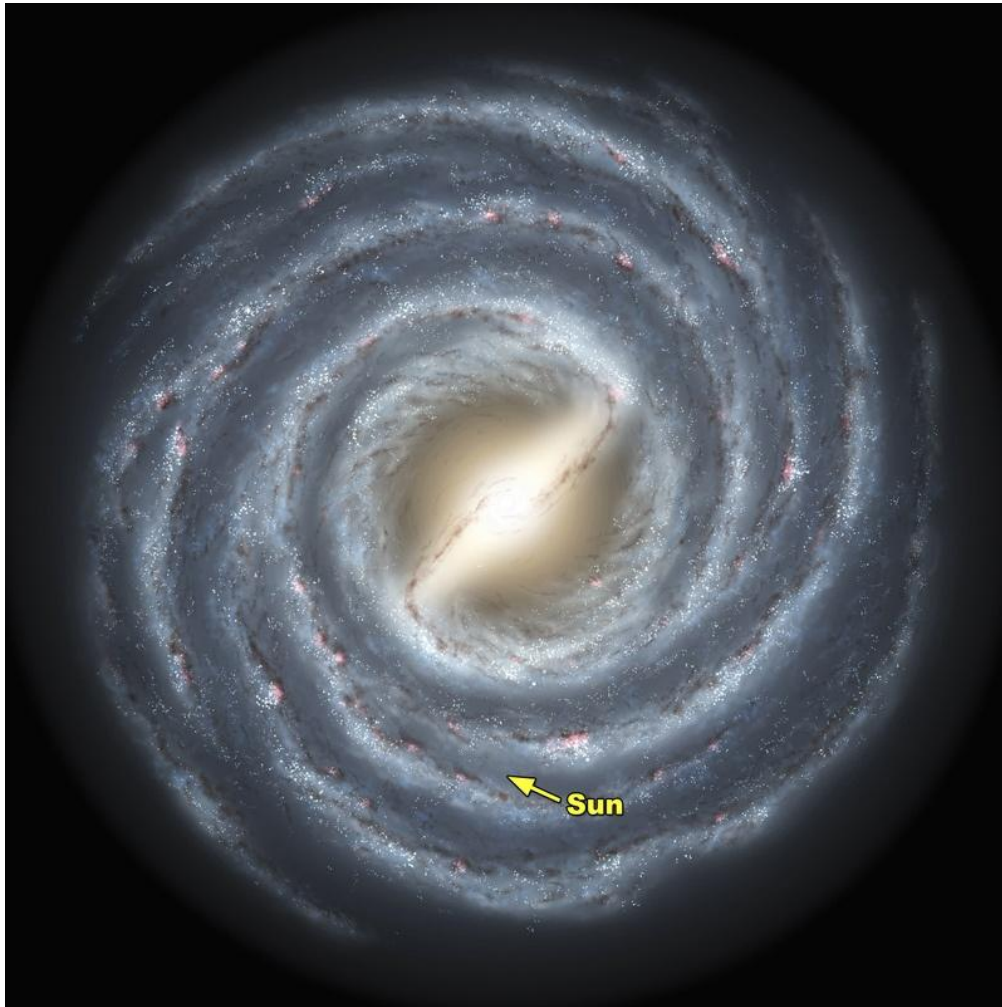


Aldebaran



The location of the Solar System in the Milky way Galaxy

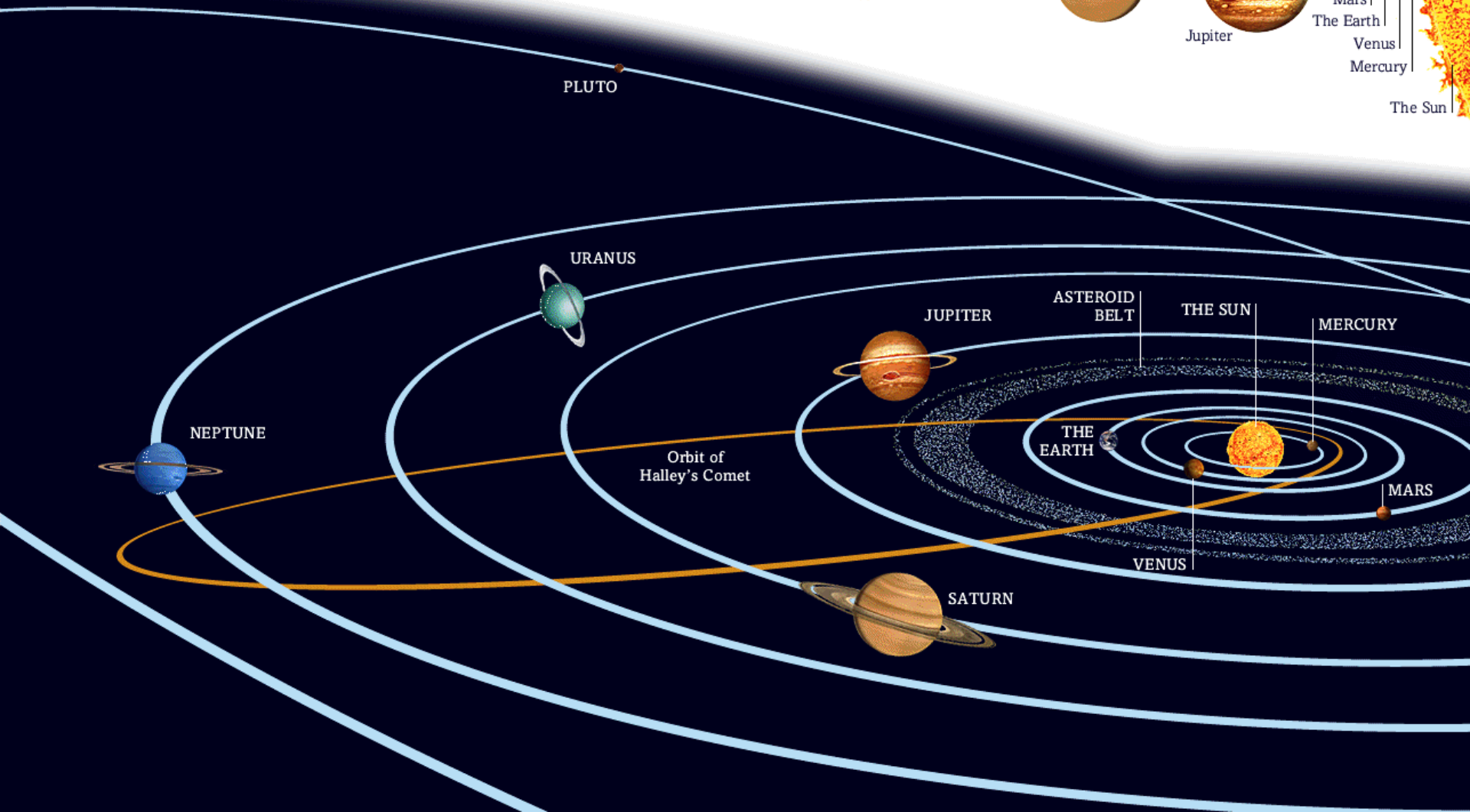
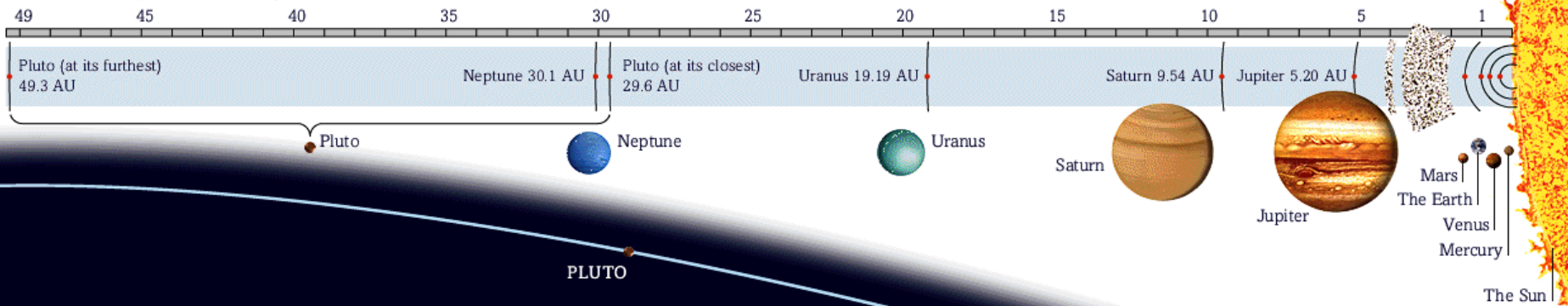
Milky way Galaxy diameter = 100,000 ly



Milky way Galaxy depth = 2000 ly

ASTRONOMICAL UNITS

(1 AU = the average distance between the Earth and the Sun)



The Sun

Our sun is a star located in our Solar System. It is a huge, spinning ball of hot gas that lights up the Earth and provides us with heat.



- *The sun is the biggest, brightest, and hottest object in the solar system.*
- *The sun is an ordinary star.*
- *Sun converts hydrogen to helium in its core*
- *Differential rotation*
 - *equator the surface rotates once every 25.4 days*
 - *near the poles it's as much as 36 days*

Basic Structure

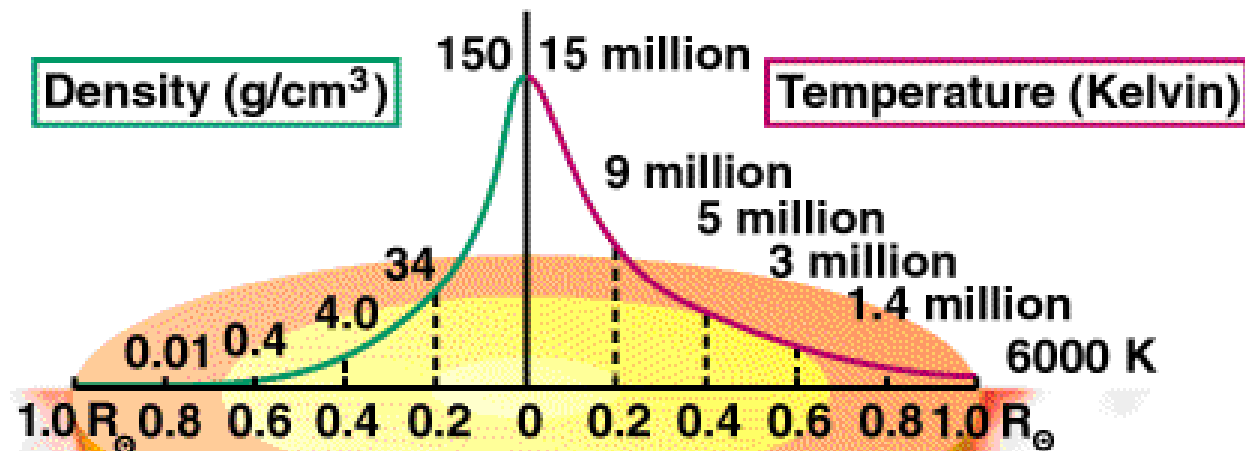
- *Average Density = 1.4 g/cm^3*
- *Core Temperature = $15,000,000 \text{ K}$*
- *Mass = $2. \times 10^{30} \text{ ton gas}$*
- *Average distance from earth $149,600,000 \text{ km}$*
- *pressure is 10 billion atmospheres in centre*
- *temperature is 14 million Kelvin in centre*
- *Surface Temperature = 5750 K*

Chemical Composition of the Sun

Here is a table of the 10 most common elements in the Sun:

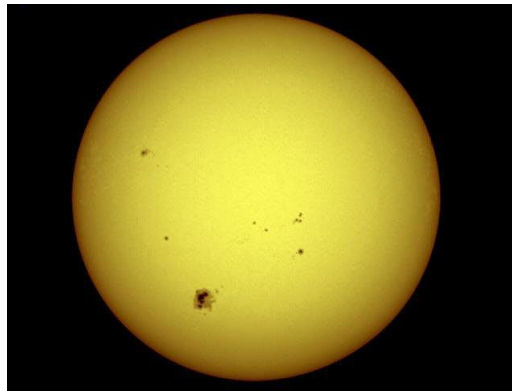
<i>Element</i>	<i>Abundance (% of total number of atoms)</i>	<i>Abundance (% of total mass)</i>
<i>Hydrogen</i>	<i>91.2</i>	<i>71.0</i>
<i>Helium</i>	<i>8.7</i>	<i>27.1</i>
<i>Oxygen</i>	<i>0.078</i>	<i>0.97</i>
<i>Carbon</i>	<i>0.043</i>	<i>0.40</i>
<i>Nitrogen</i>	<i>0.0088</i>	<i>0.096</i>
<i>Silicon</i>	<i>0.0045</i>	<i>0.099</i>
<i>Magnesium</i>	<i>0.0038</i>	<i>0.076</i>
<i>Neon</i>	<i>0.0035</i>	<i>0.058</i>
<i>Iron</i>	<i>0.0030</i>	<i>0.014</i>
<i>Sulfur</i>	<i>0.0015</i>	<i>0.040</i>

How Temperature and Density Vary Inside the Sun



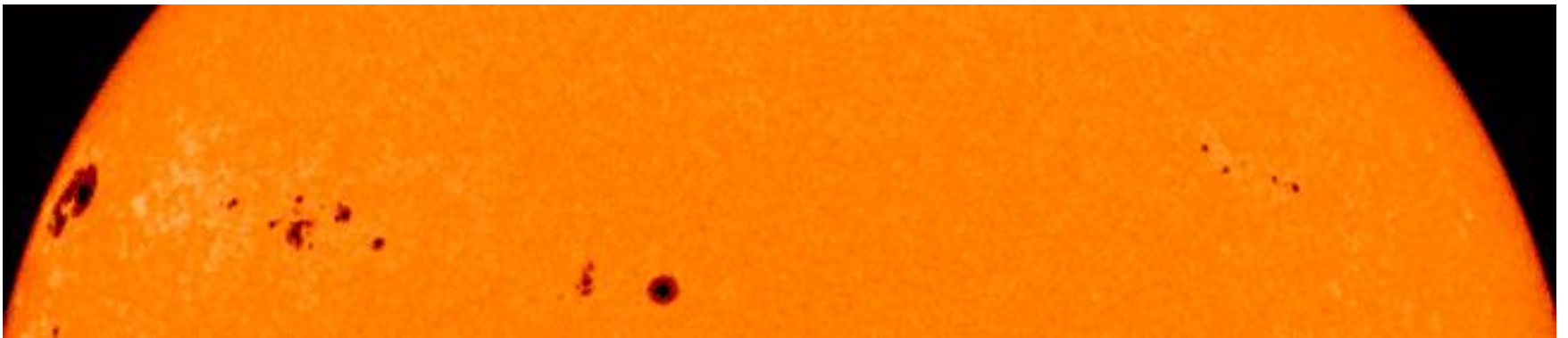
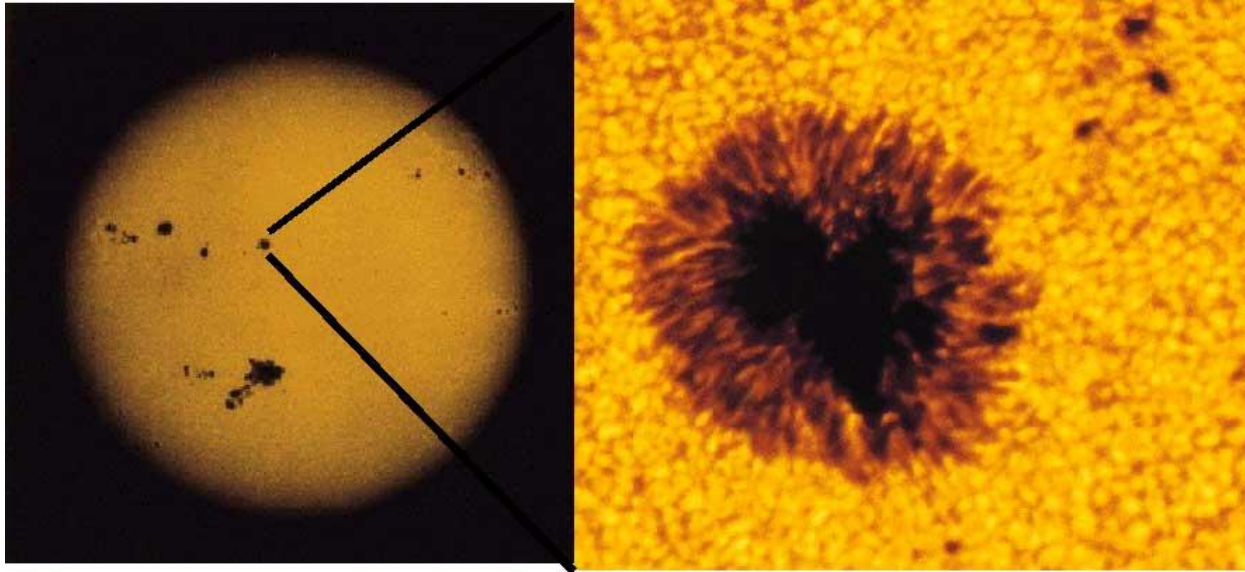
Solar Characteristic

Photosphere: "sphere of light", the visible surface of the Sun.

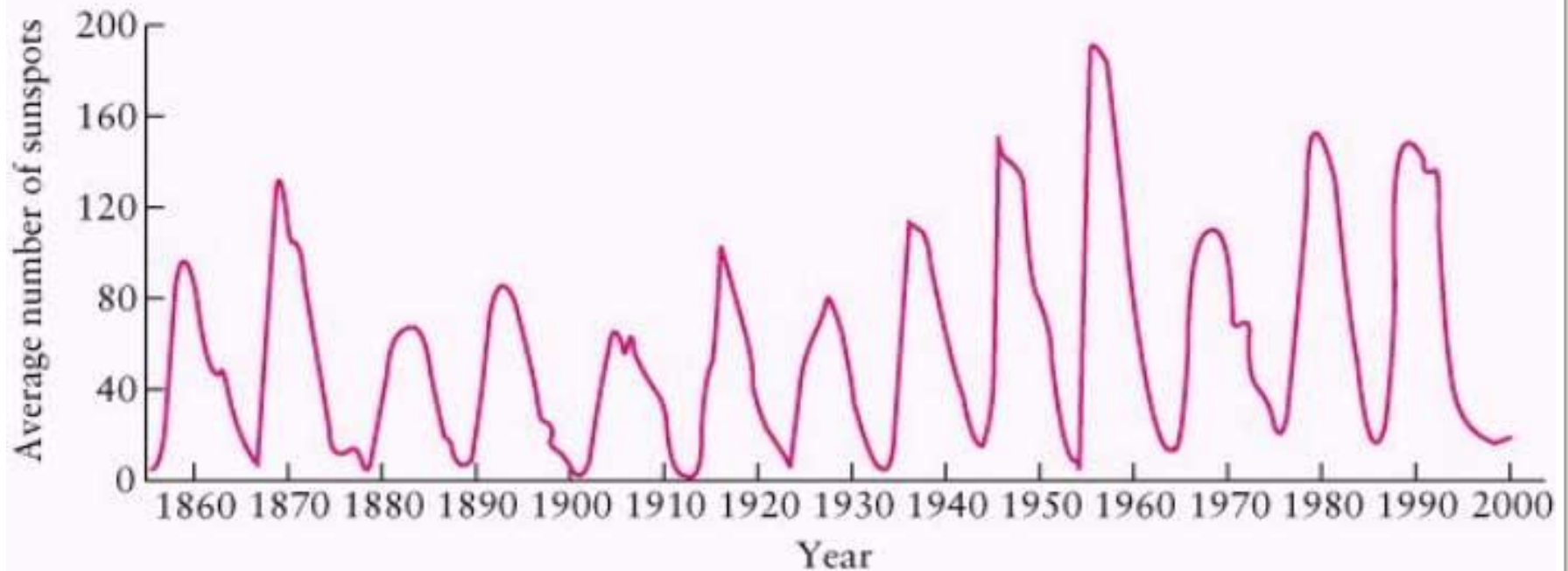


$$T = 5750 \text{ K}$$

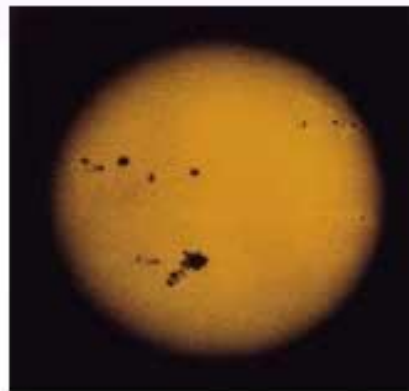
SUNSPOTS



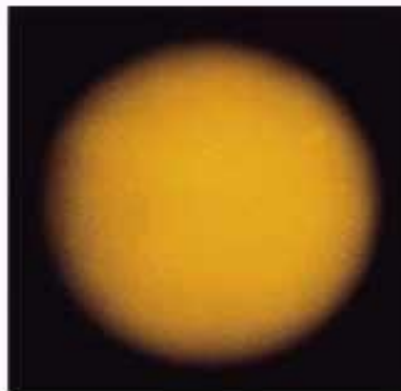
The annual change in numbers of **sunspots** reveals that the Sun experiences an 11-year solar cycle.

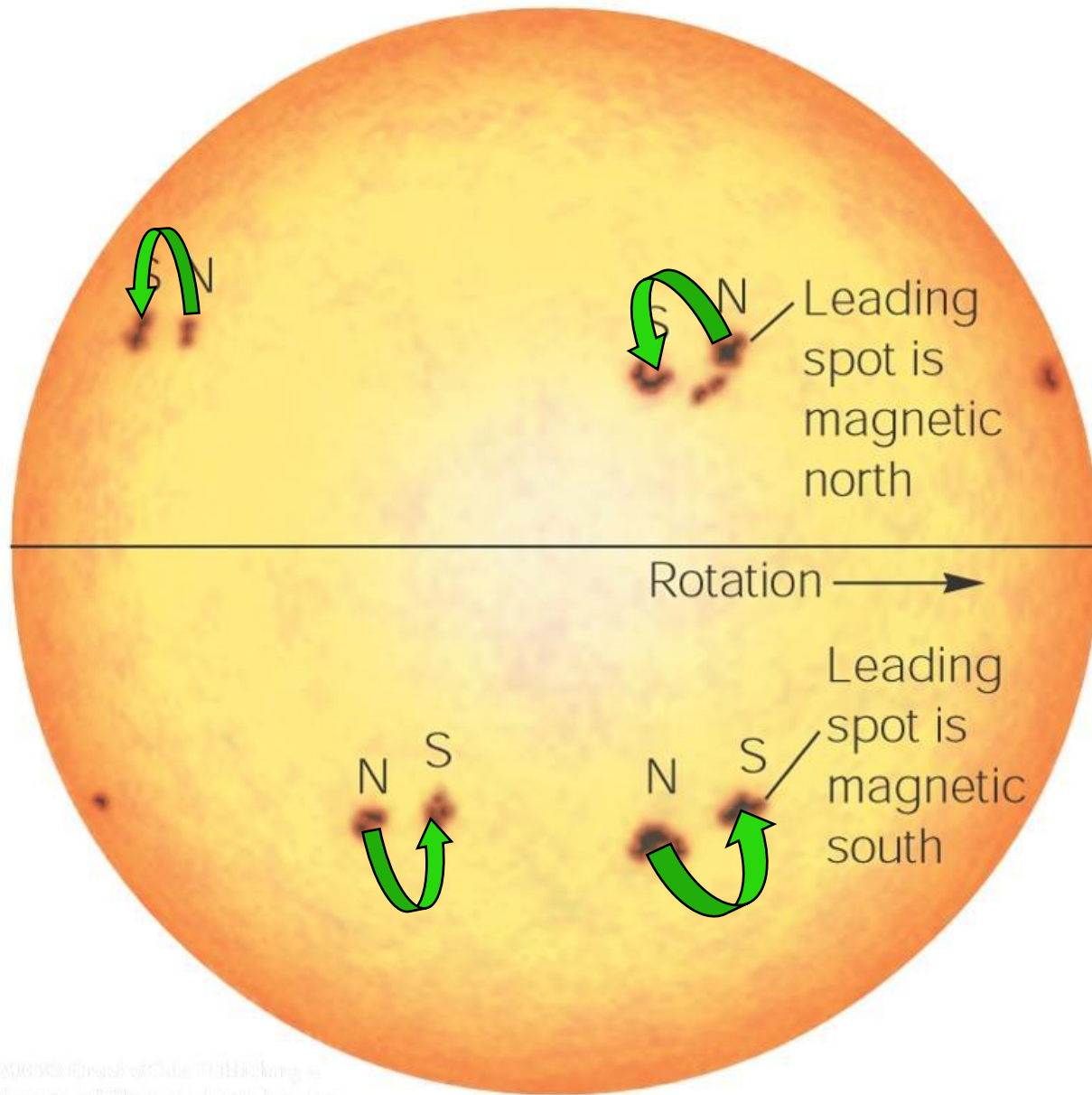


*Maximum
number*

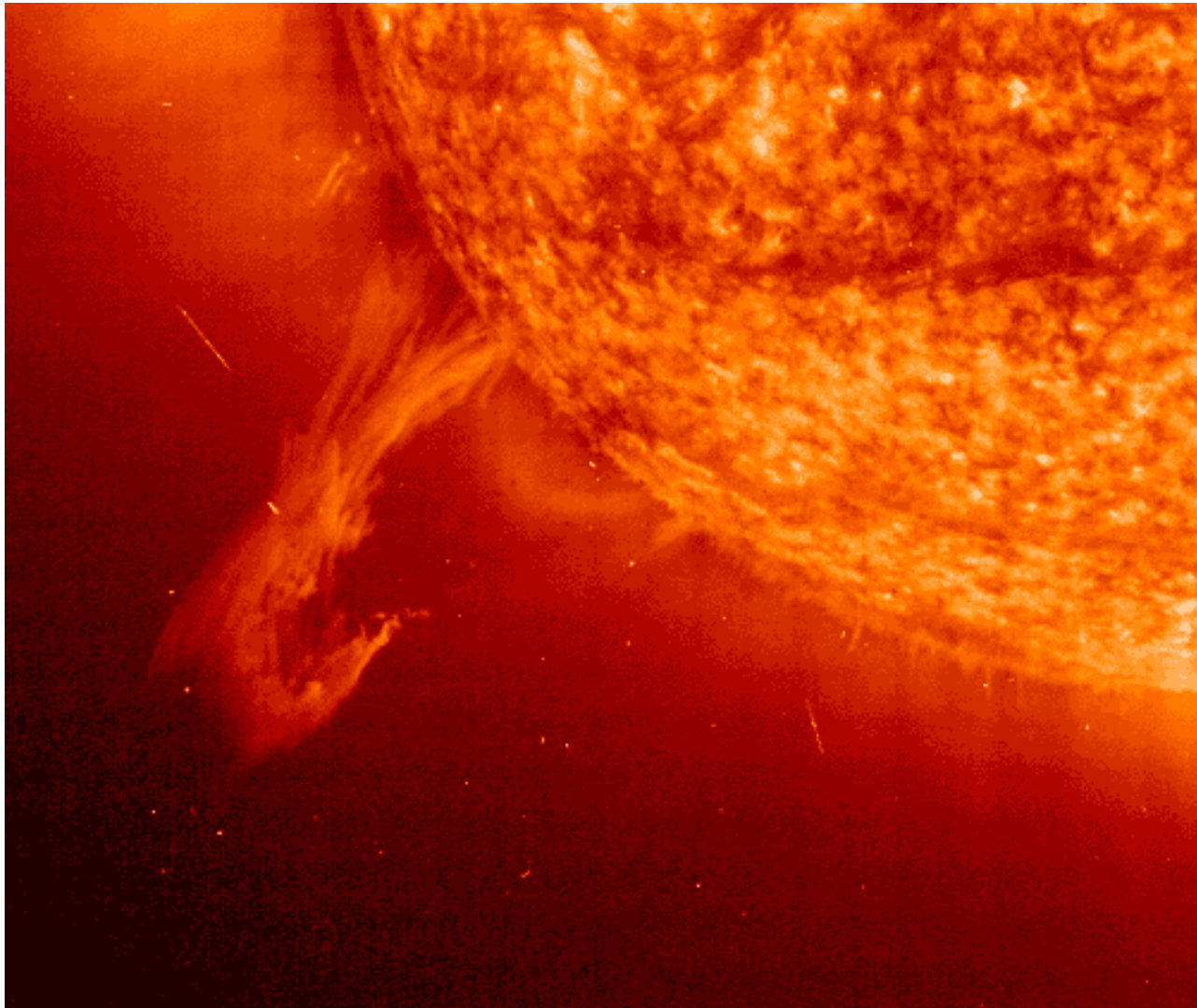


*Minimum
number*





Prominence



Chromosphere: "sphere of color", visible during solar eclipses

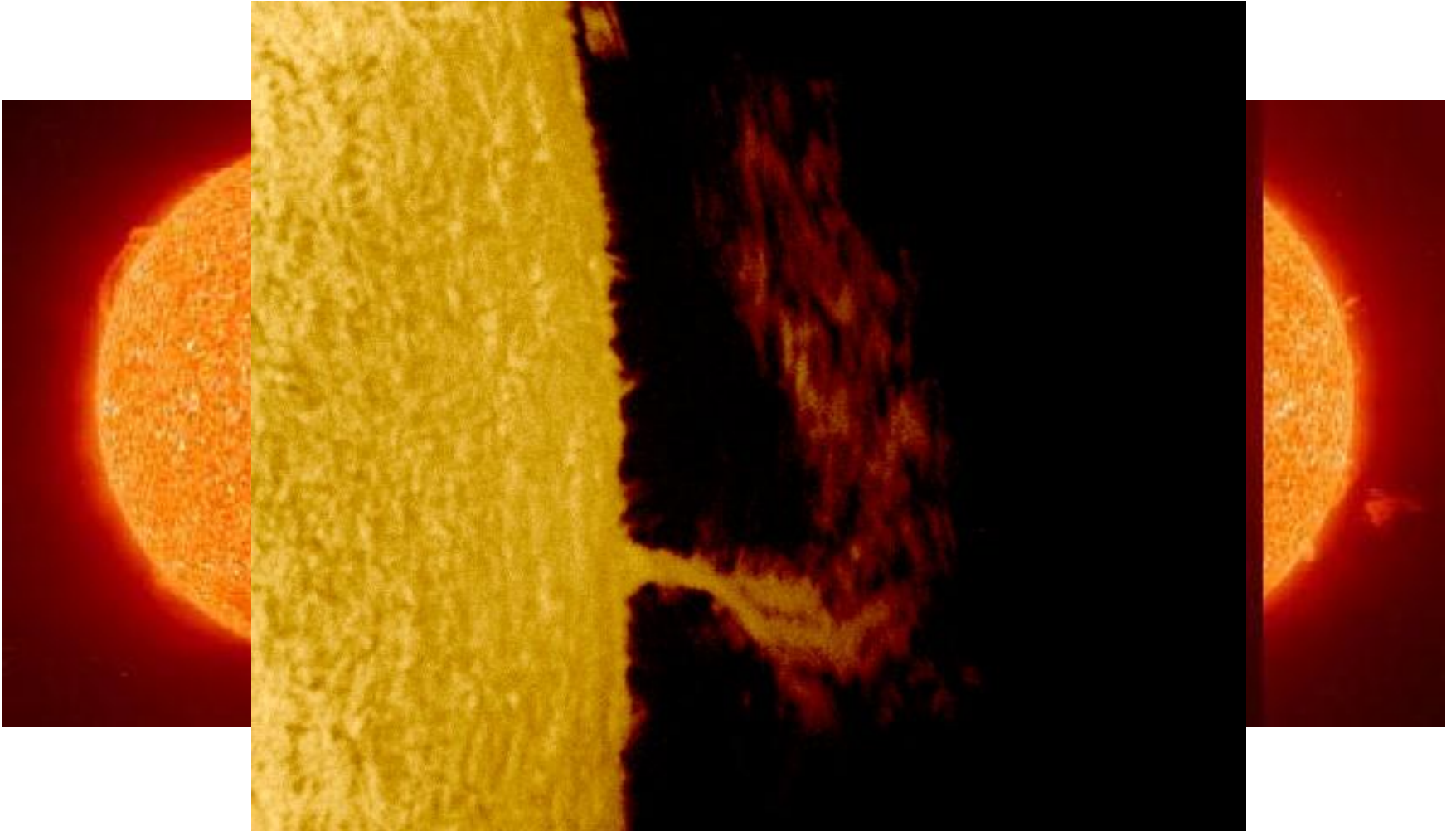


just outside of photosphere

hot gas

$T = 10,000\text{ K}$

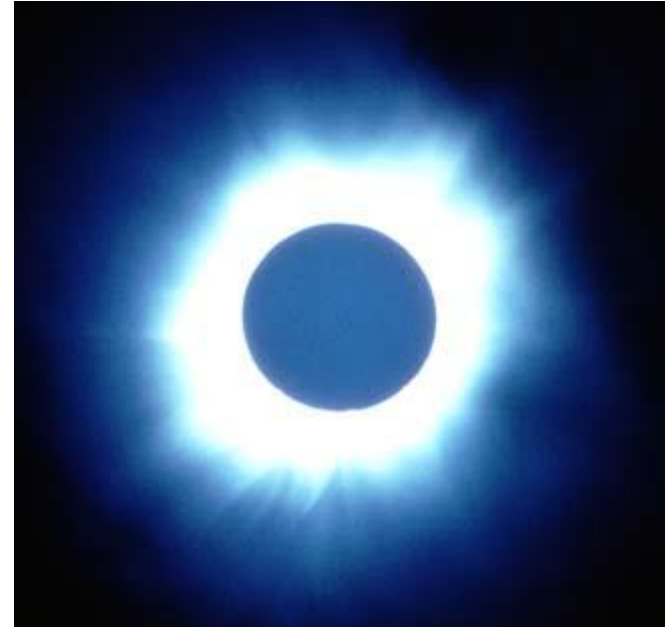
Solar Flare



***Corona** - the Sun's outermost atmosphere.*

*The outflow of gas in this region is called the **solar wind**, which is protons and electrons that have escaped the Sun's gravity.*

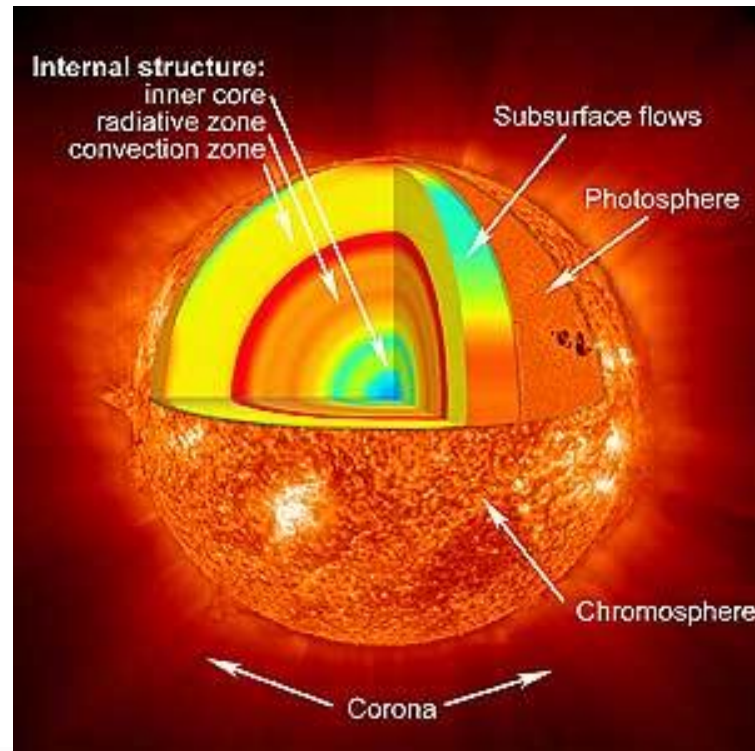
Very hot (few million degrees) visible during eclipse.



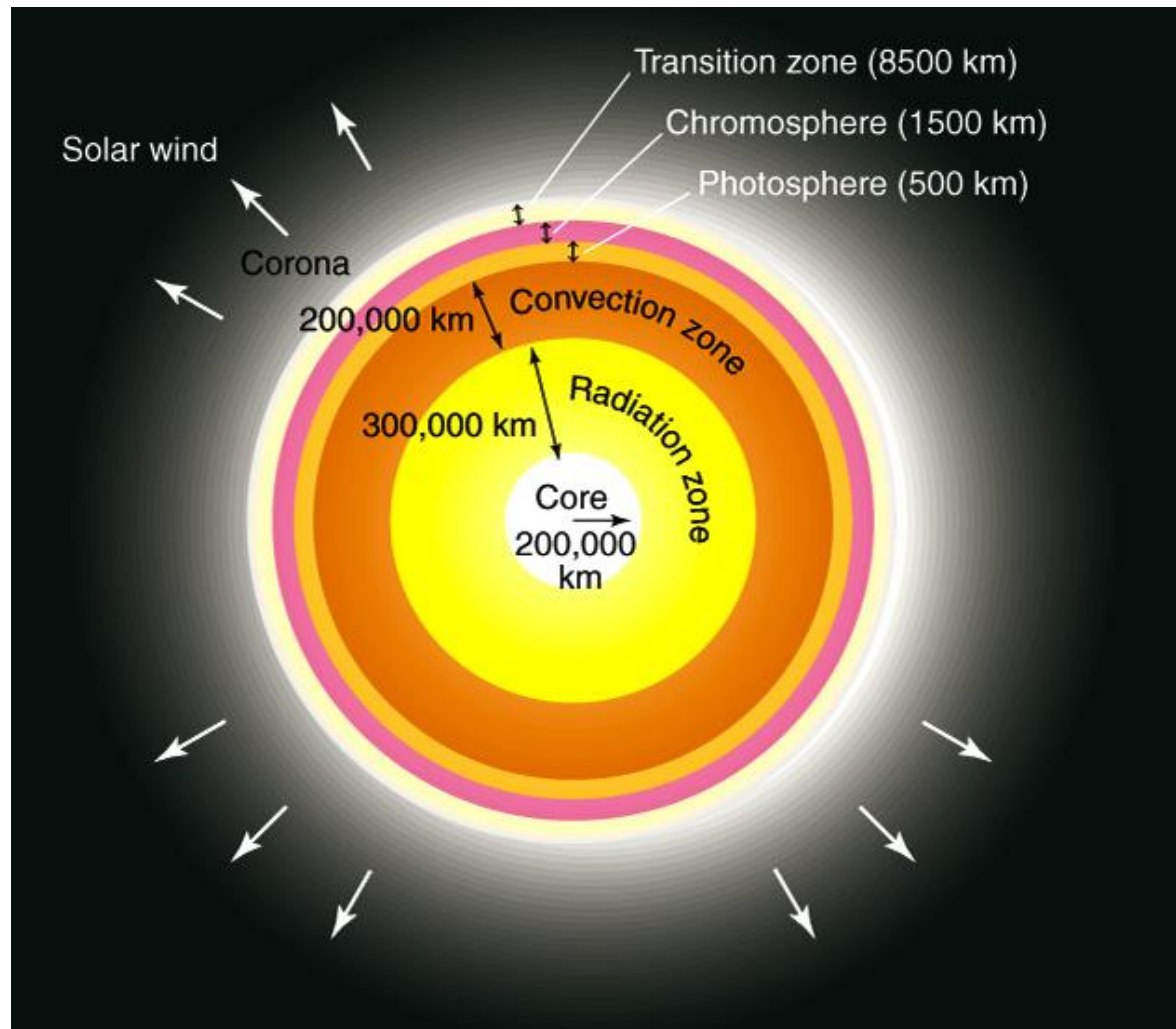
The sun's corona, or extended outer layer, is a region of plasma that is heated to over 1000000 celcius. the mean velocity of these particles is about 145km/s, which is below the solar escape velocity of 618km/s.

The Sun's Interior

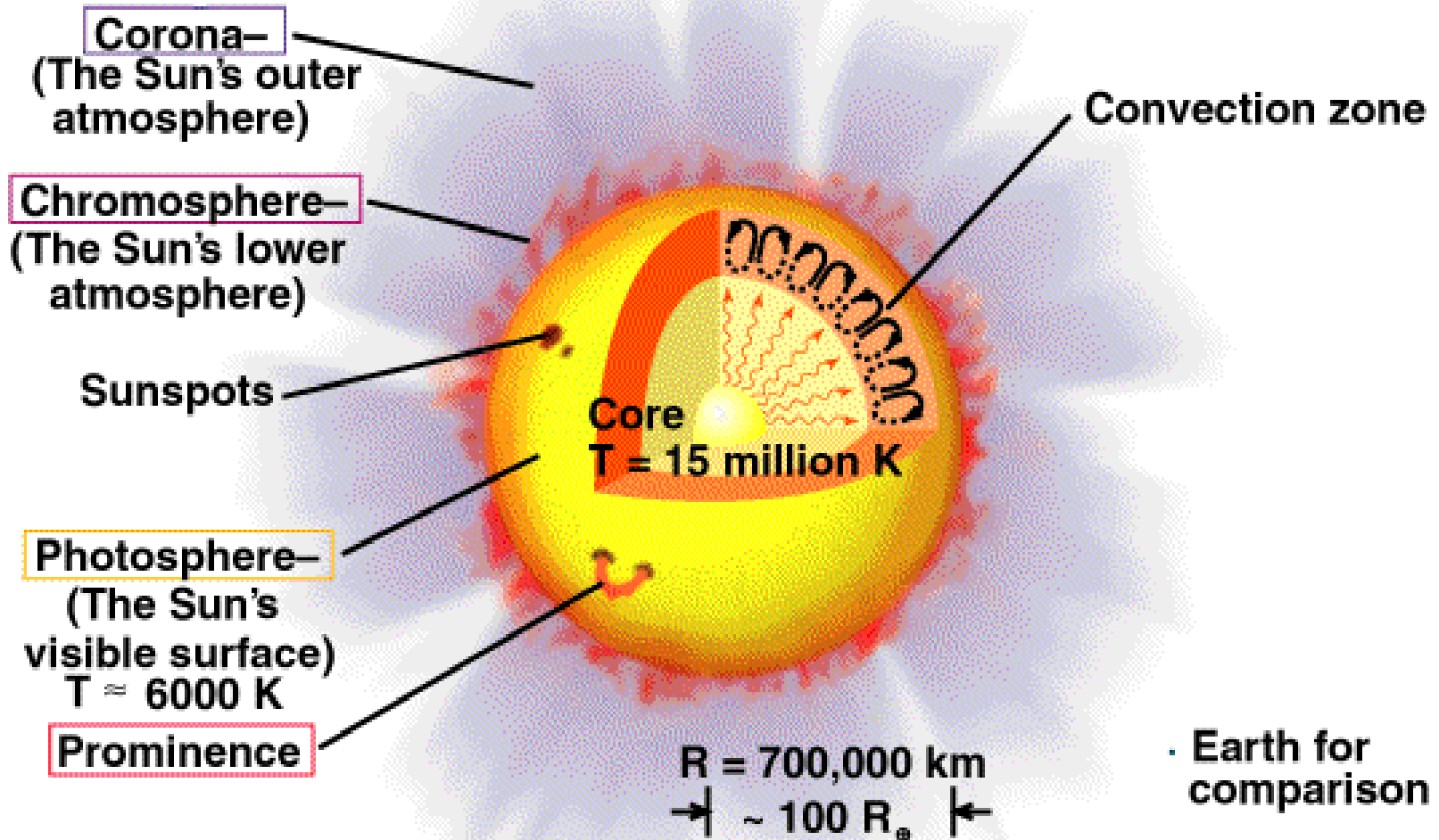
- *Radiative zone - a region inside a star where energy is transported outward by the movement of photons.*
- *Convective zone - a layer inside a star where energy is transported outward by means of heat flow through the gasses of the star (convection).*



Layers of the Sun



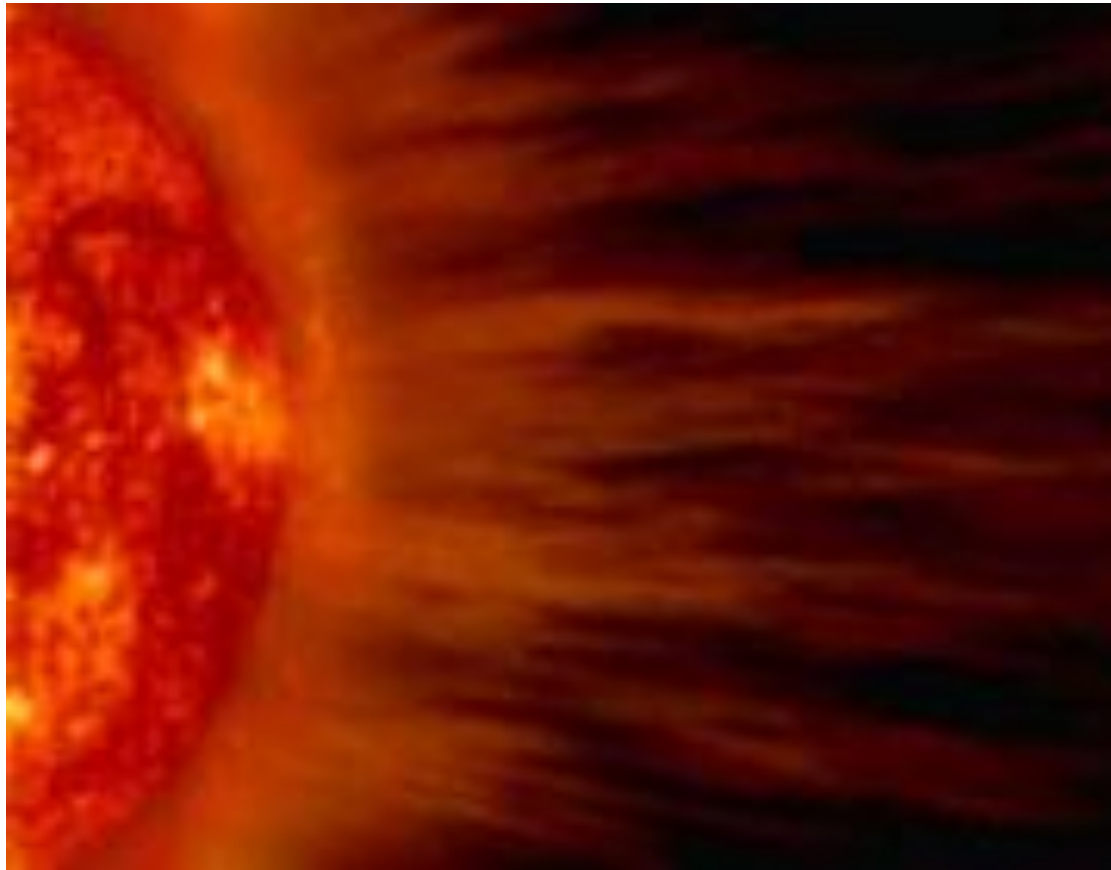
Cut Away View of the Sun



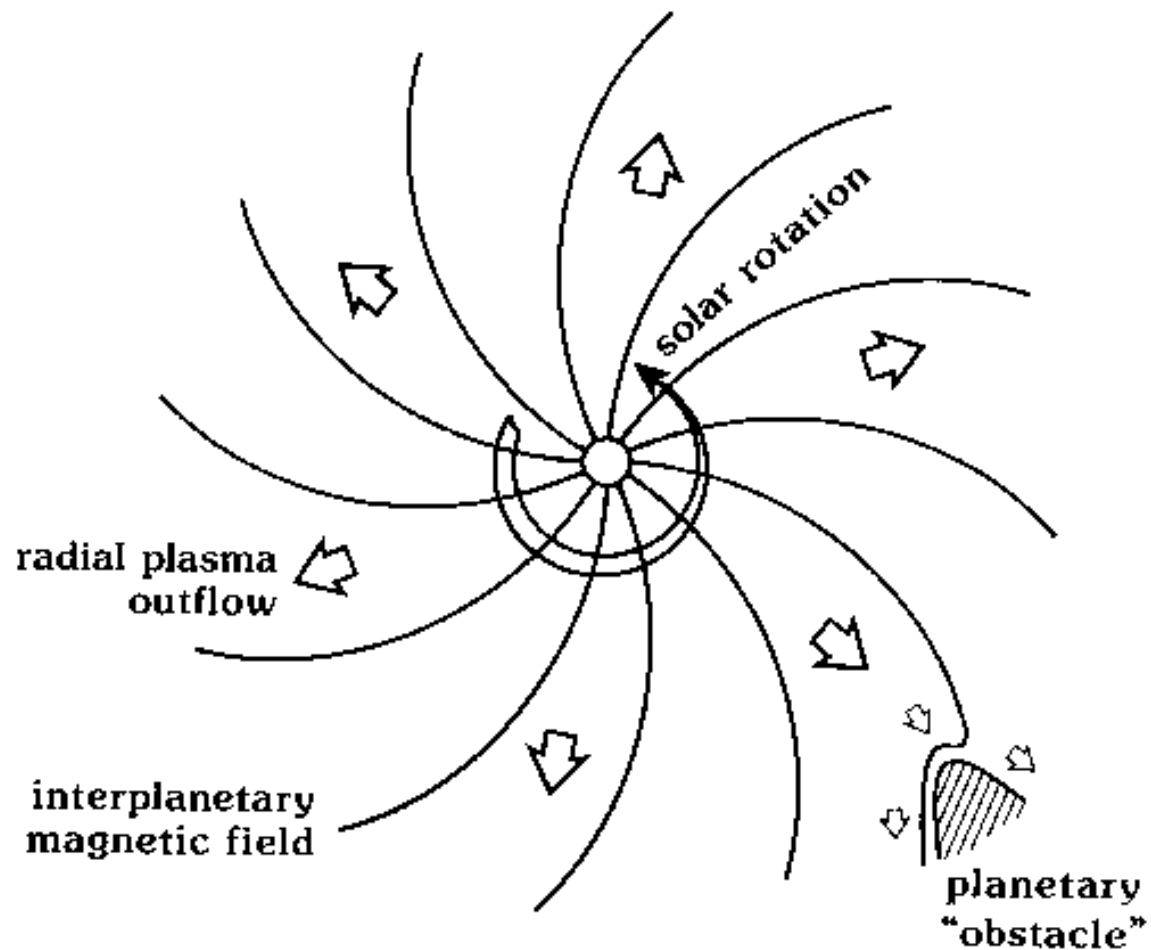
The Solar Corona: Source of the solar wind

The solar wind is a stream of charged particles-plasma-ejected from the upper atmosphere of the sun.

It contains electrons & protons with energies of about 1kv



The Solar Wind and Interplanetary Magnetic Field



Fusion of Hydrogen into Helium

The Sun's Source of Energy

- ***Nuclear Reactions:*** *The Sun's energy comes from nuclear reactions in its core.*

More specifically, the energy comes from the fusion (joining) of nuclei of hydrogen atoms into nuclei of helium atoms. The actual fusion process takes place in several steps, which may be summarized as follows:

Four hydrogen nuclei get fused into one helium nucleus, accompanied by the emission of neutrinos and release of energy:



^1H = nucleus of hydrogen atom

^4He = nucleus of helium atom

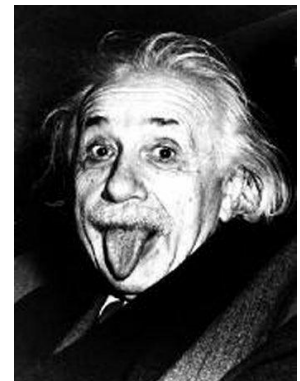
Neutrinos are subatomic particles that travel close to the speed of light and only rarely interact with matter.

Where Does the Energy Come from in the Fusion of Hydrogen into Helium?

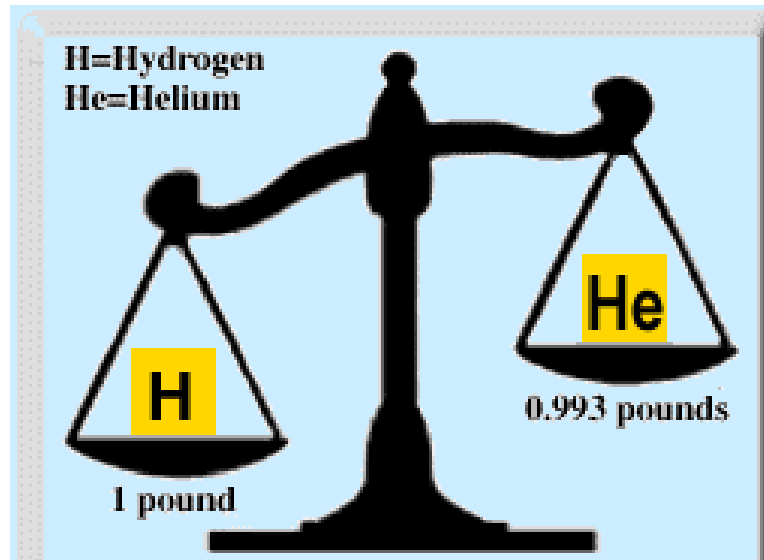
During the fusion of hydrogen, approximately 0.7% of the mass of hydrogen is converted into energy. This means that the resulting helium has 0.7% less mass than the original hydrogen. Einstein's famous formula tells us how much energy this loss of mass generates:

Hydrogen + Hydrogen = Helium + energy
difference in mass converted to energy

famous Einstein $E=mc^2$
mass=energy



Since the speed of light (c) is very large and comes in squared, the amount of energy generated by the fusion of hydrogen is very large, even though the amount of mass that disappears is very small.



If we could weigh the helium produced by the fusion of 1 pound of hydrogen, we'd find it weighs only 0.993 pounds. The difference, namely 0.007 pounds, has been converted into energy.

NUCLEAR "ENERGY" is the conversion of mass to energy