

فیزیک جو

درس اول

صحرائی

گروه فیزیک دانشگاه رازی

References:

منابع:

1. An Introduction to Atmospheric Physics, Fleagle & Businger
2. The Upper Atmosphere Meteorology and Physics, Craig.
3. Internet
4. هواشناسی عمومی – دکتر قائمی – سمت.
5. مبانی هواشناسی – دکتر قائمی.
6. هواشناسی فیزیکی – رتالاک – ترجمه صادقی

Assessment

- **Mid semester exam (Sun. 6 Azzar.):**
30%
- **Final exam:** **70%**

Research Topics

برهمکنش باد خورشیدی و شهاب

لایه ازن

لایه بار الکتریکی

جو زمین

ماهواره مصنوعی

تعیین چگالی اتمسفر

توفان مغناطیسی زمین

شفق قطبی

تخریب لایه ازن

ارتفاع سنجی تولید و اتلاف در یونسفر

لایه هموسفر

شهاب سنگ

اتموسفر

ترکیبات اتمسفر

میدان الکتریکی یونسفر

جذب تابش در اتمسفر

مسیر حرکت سیارات

اصطلاحات رایج مکانیک سماوی

اتموسفر

جوسنج

پلاسماپوز

ترکیب مجدد در یونسفر

ساختار اتمسفر

فروغ آسمانی

منشا تابش سیاره ای

انتقال و پخش در یونسفر لایه یونسفر

فیزیک جو

لایه اکزوسفر

ضخامت نوری

تونل باد

محفظه مغناطیسی پلاسماسفر

علم بالستیک بارش شفق قطبی

سنجش دمای اتمسفر

انتشار و جذب فروغ آسمانی شار فوتونی

توفان مغناطیسی

لایه تروپوسفر

قوانین کیلر

بحران لایه ازن

اتموسفر و تابش خورشیدی

تعادل ترمودینامیکی

لایه ماگنتوسفر

میدان مغناطیسی زمین

ماگنتوسفر و باد خورشیدی

پلاسماسفر

جزیره مغناطیسی

یونش فوتوالکترونی اتمسفر

باد خورشیدی

کمر بند تشعشعی زمین

انتقال و پخش یون در پلازما

پیکربندی شفق قطبی

چگالی الکترونی یونسفر جذب تابش در اتمسفر سوراخ لایه ازن تولید تابش در اتمسفر

- (1) کلیاتی در باره ساختار و خواص جو
- (2) نگاهی کوتاه به مساله تابش و رژیم گرمائی.
- (3) اثرات گرانش
- (4) ترکیبات آرام کره و میان کره، ازن جوی
- (5) ترکیب و ساختار گرم سپهر
- (6) یون سپهر و نواحی آن
- (7) مقدمه ای بر میدان مغناطیسی زمین و تغییرات آن
- (8) مقدمه ای بر درخشندگی جو و نورهای قطبی



What is atmospheric physics?

Atmospheric physics is the application of physics to the study of the atmosphere.

- **all the bits of traditional physics that are needed to study the behaviour of the atmosphere**



What is the atmosphere?

**It's a thin layer of gases surrounding our planet.
The atmosphere is held to the planet by the force of gravity**

Earth's Blanket

The atmosphere is the gaseous envelope that surrounds the earth and constitutes the transition between its surface and the vacuum of space.

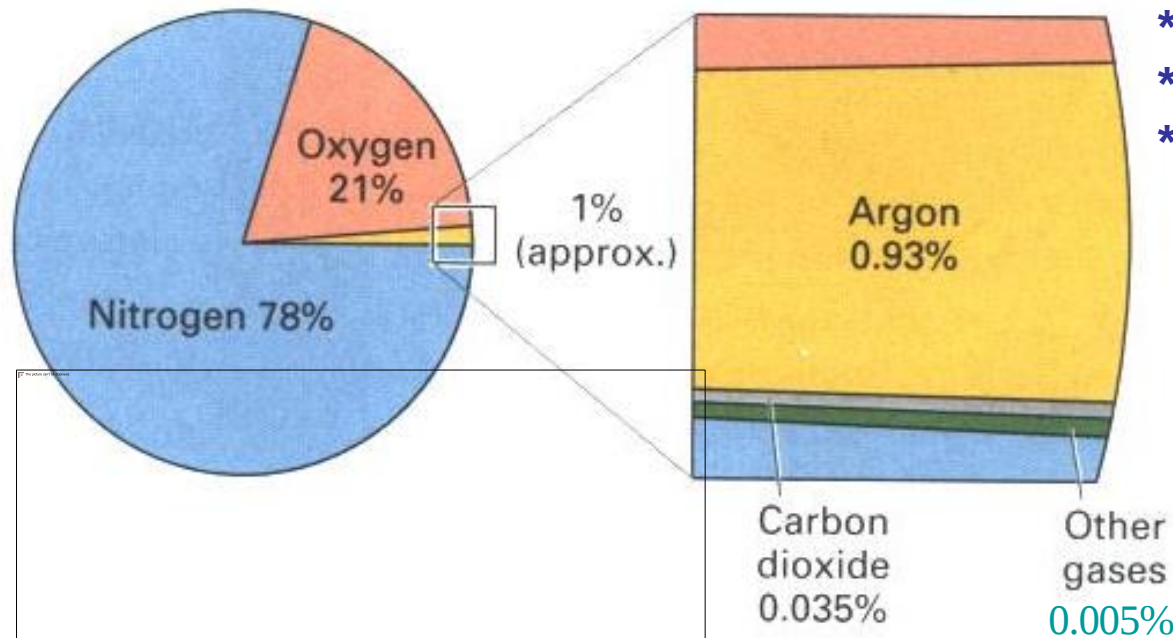
The atmosphere is composed primarily of nitrogen (N_2) and oxygen (O_2) and is made up of many layers of air, each one identified by their thermal characteristics or temperature changes, chemical composition, movement and density.

Life on earth is supported by this atmosphere, solar energy, and our planet's magnetic fields.



Atmospheric Composition

Values show percentage by volume for dry air. Nitrogen and oxygen form 99 percent of our air, with other gases, principally argon and carbon dioxide, accounting for the final 1 percent.



*Neon-----0.002%
*Helium-----0.0005%
*Krypton-----0.0001%
*Hydrogen-----0.00005%

VARIABLE gases in the atmosphere and typical percentage values are:

*Water vapor-----0 to 4%
*Carbon Dioxide----0.035%
*Methane-----0.0002%
*Ozone-----0.000004%

THERMAL STRUCTURE OF THE ATMOSPHERE



According to temperature, the atmosphere contains five different layers:

Troposphere

زیرین کره

Stratosphere

آرام کره

Mesosphere

میان کره

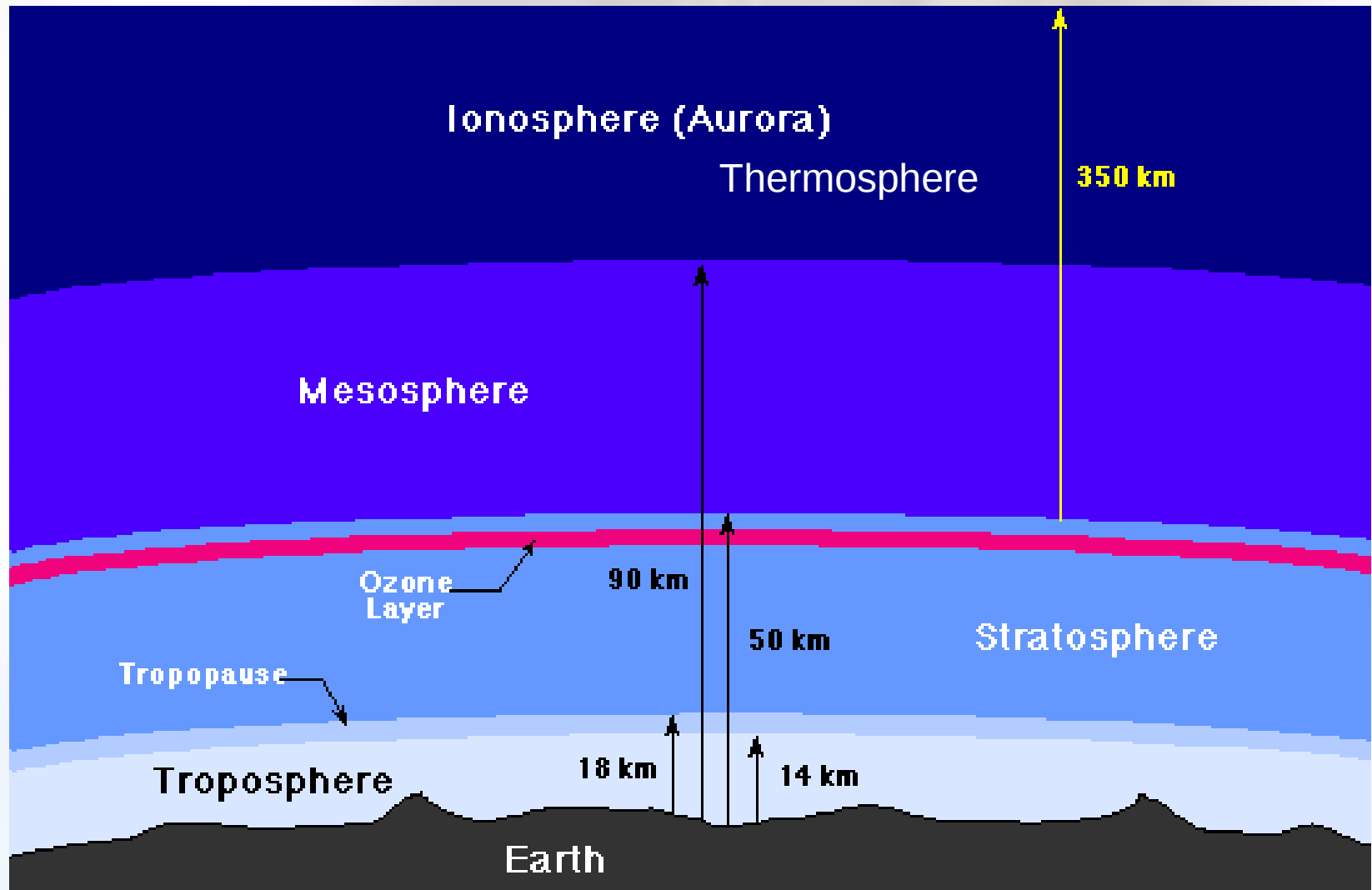
Thermosphere

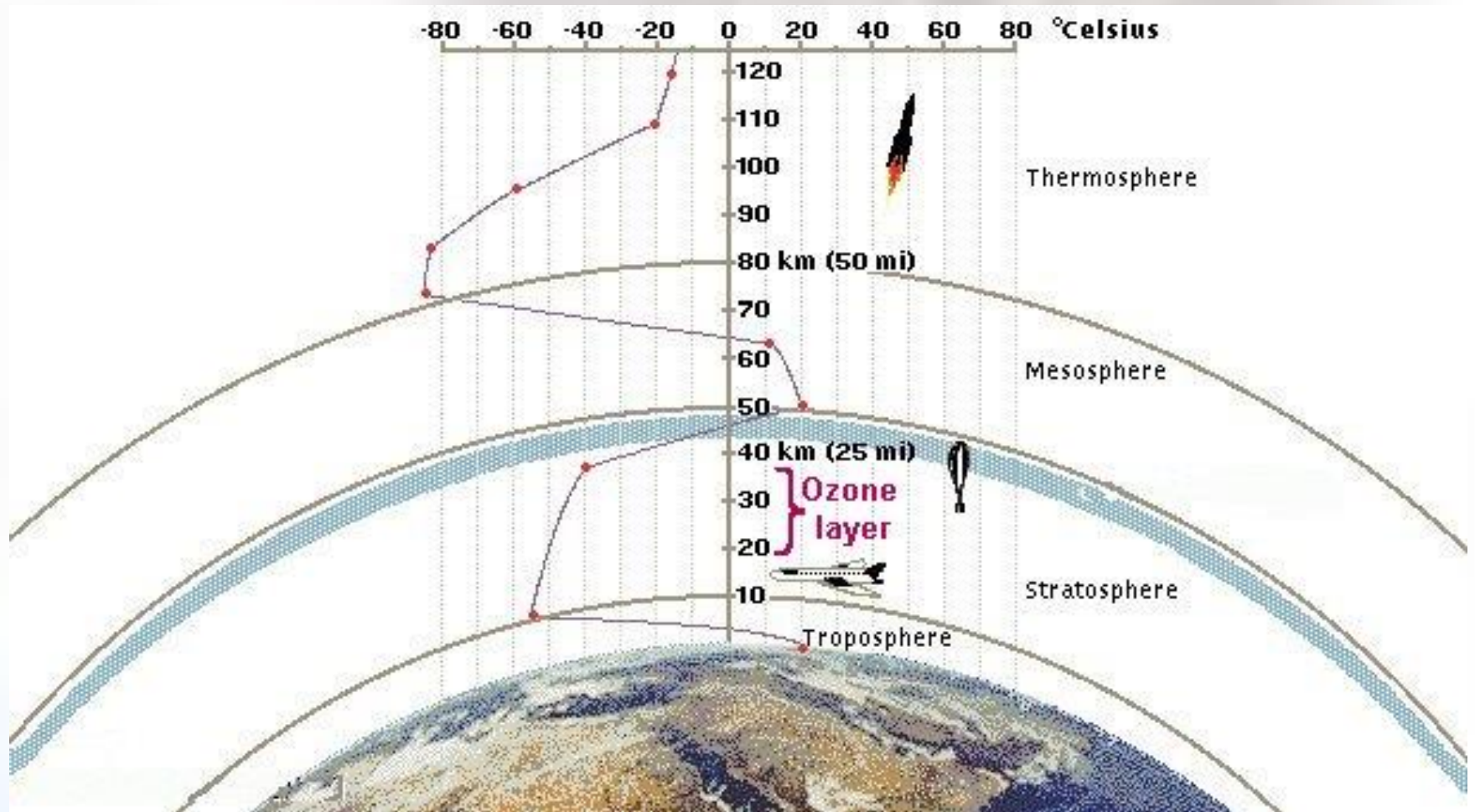
گرم کره

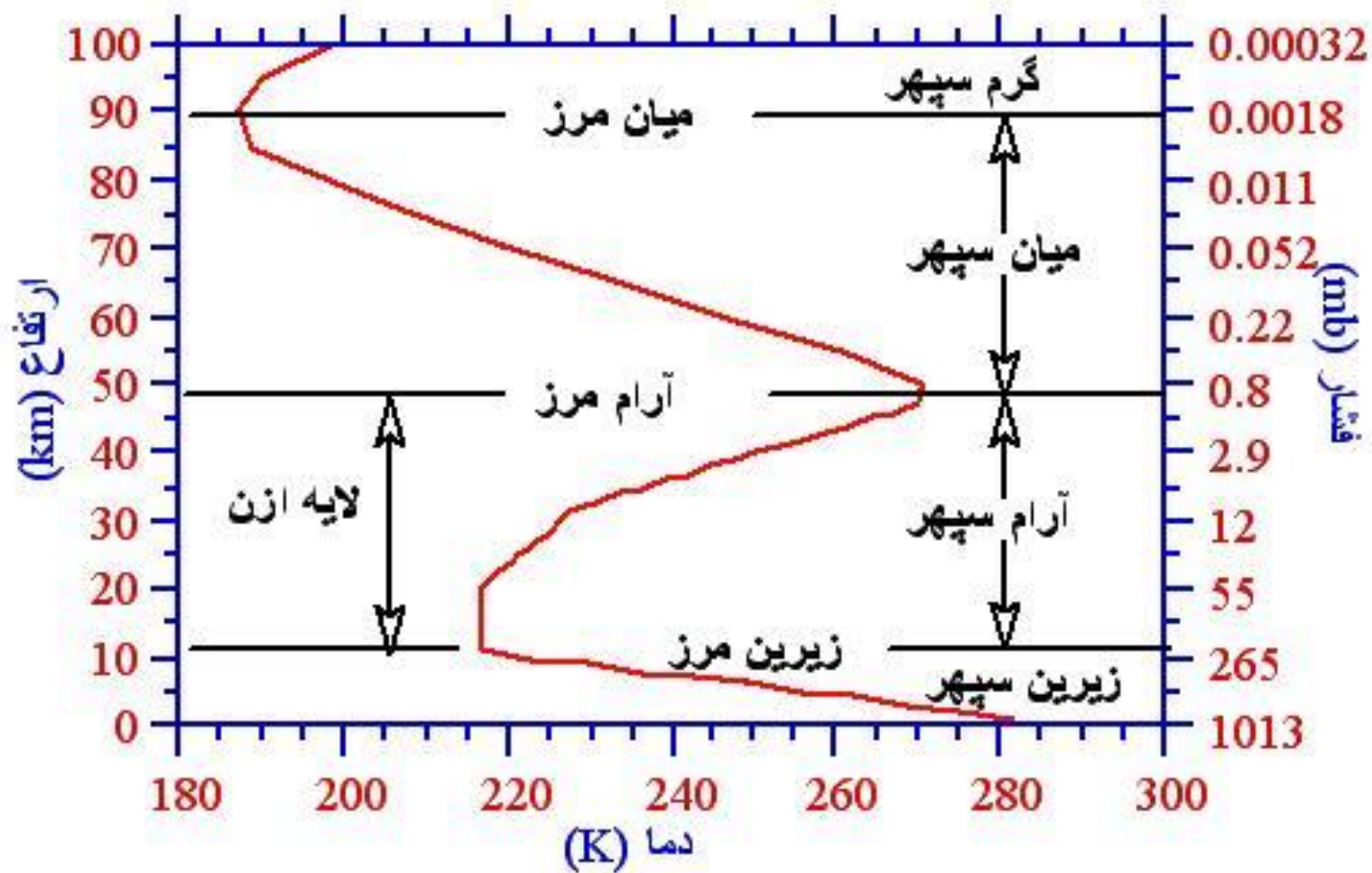
Exosphere

برونکره

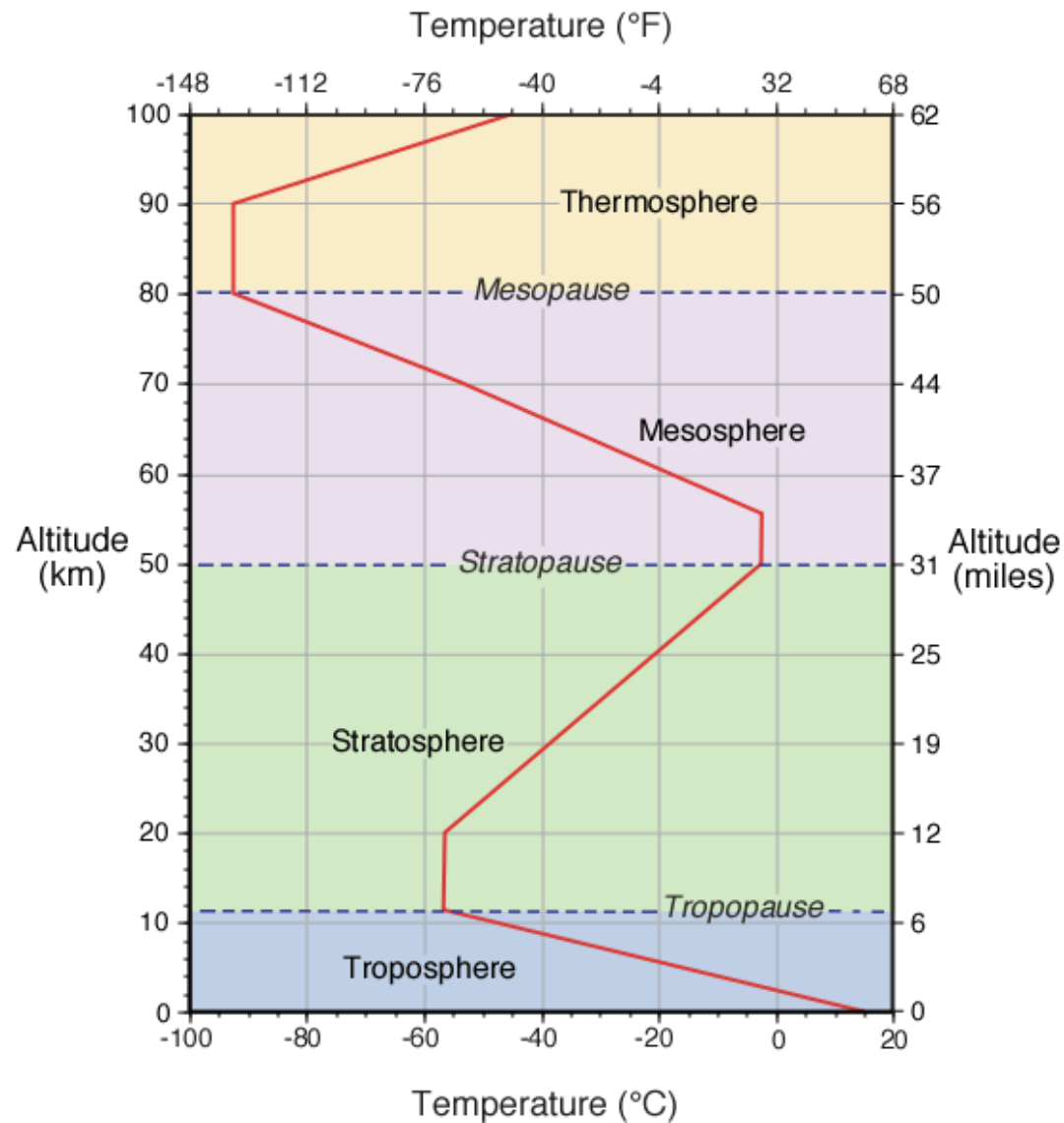
THERMAL STRUCTURE OF THE ATMOSPHERE







THERMAL STRUCTURE OF THE ATMOSPHERE



Layer One (Troposphere)

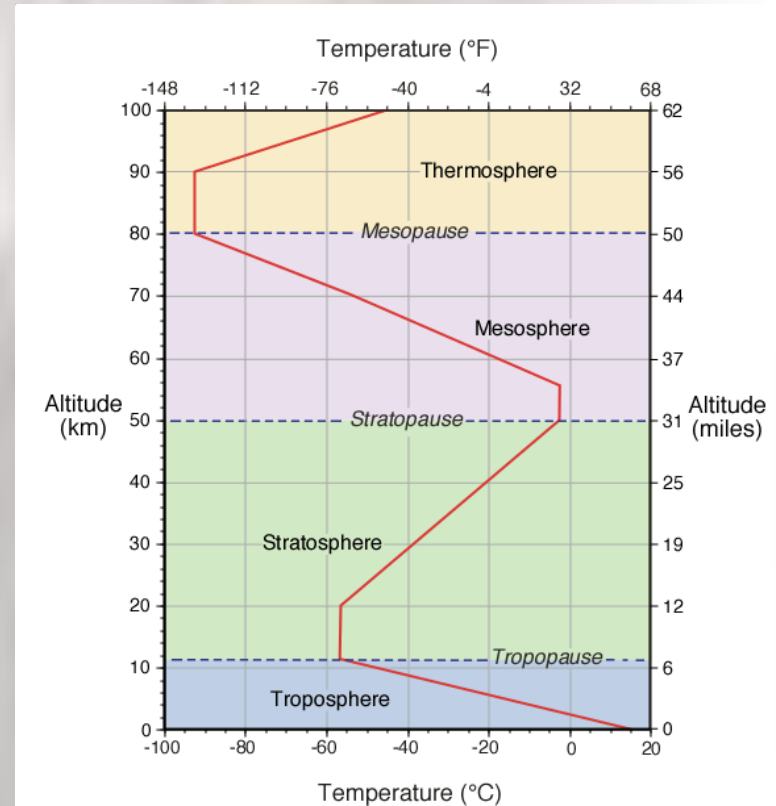
The layer closest to the Earth

Often referred to as “The Weather Layer”

Rain snow and wind stick to this layer.

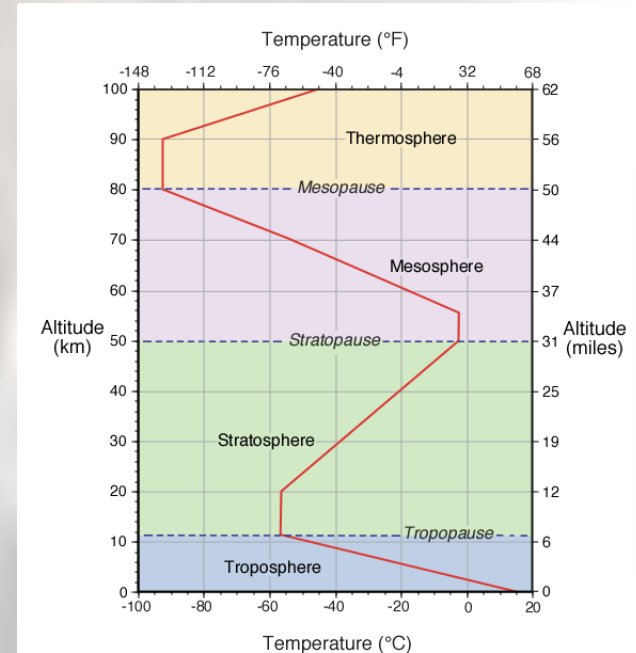
Planes fly in this layer

Below an altitude of 12 km the temperature decreases from 15 °C to -55 °C as altitude increases.



Layer Two (Stratosphere)

- In the stratosphere (12 km - 50 km) the temperature increases with altitude from -55 °C to -2 °C.
- Ozone layer found in the stratosphere
- Ozone is a gas that absorbs harmful UV rays and protects us from too much solar radiation. Pollution has created a hole in the Ozone layer over the South Pole.



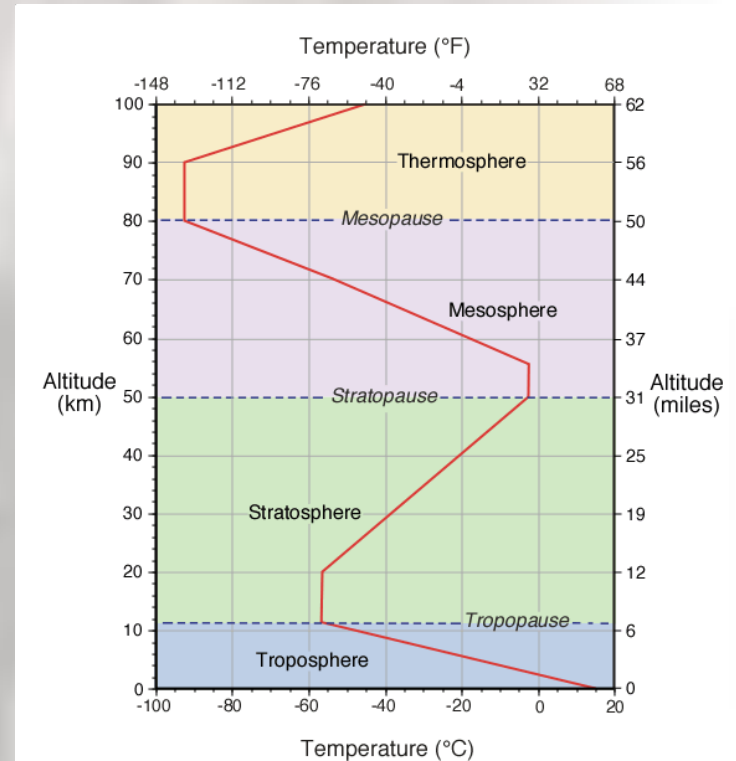
Layer Three (Mesosphere)

- In the mesosphere (50 km - 80 km) the temperature decreases (-2 °C to -90 °C) .

Atmosphere reaches it's coldest temperature of about -90° C.

A lot of meteors disintegrate in this layer from friction from entering the Earth's atmosphere.

The stratosphere and mesosphere together are sometimes referred to as the *middle atmosphere*.

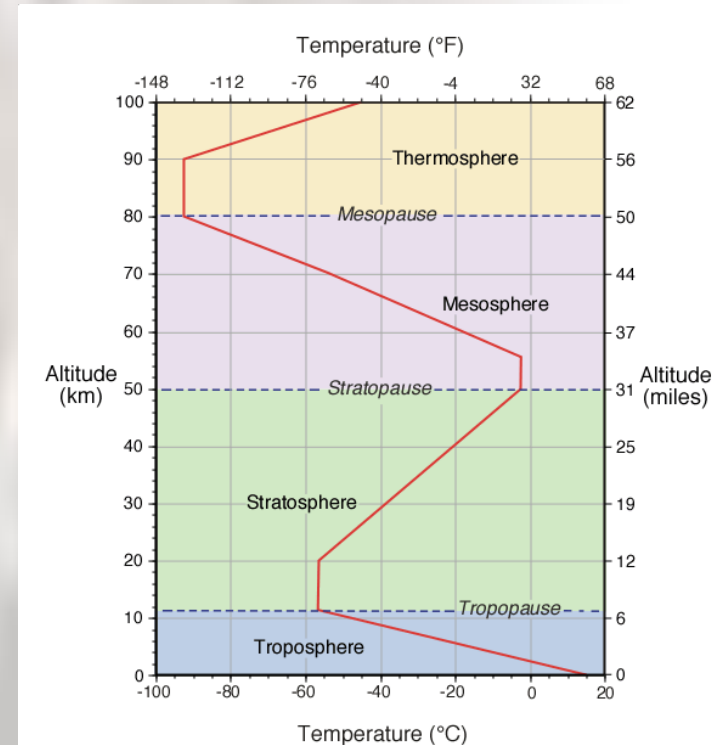


Layer Four (Thermosphere)

The thermosphere begins 80km above the earth.

Temperatures in the thermosphere go up when moving farther away from ground level due to the sun's energy.

- **Hottest of all layer, Temperatures in this layer can get as high as 1300-1800°C.**
- Includes the Ionosphere, a region of the atmosphere, which is filled with charged particles.



The thermosphere is the hottest layer in the atmosphere. In the thermosphere, gamma, X-ray, and specific wavelengths of ultraviolet radiation are absorbed by certain gases in the atmosphere. The absorbed radiation is then converted into heat energy.

The ionosphere is a region in the upper atmosphere that has a large concentration of ions and free electrons. Ions are atoms and that are electrically charged because they have either lost or gained one or more electrons. The ionosphere plays a major role in radio communications. The ionosphere reflects radio waves back to earth

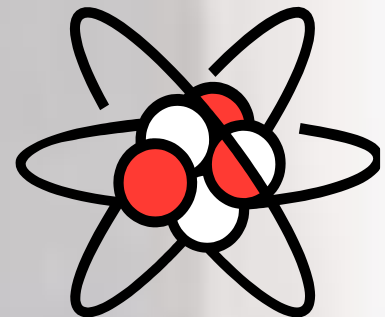
Layer Five (Exosphere)

At the top of the thermosphere, is the point at which the atmosphere thins and atoms and molecules begin to shoot off into space.

Not stable- gas molecules can escape into space

The exosphere is the highest layer of the atmosphere, 900 km above the Earth.

The thermosphere and the exosphere together make up the upper atmosphere.

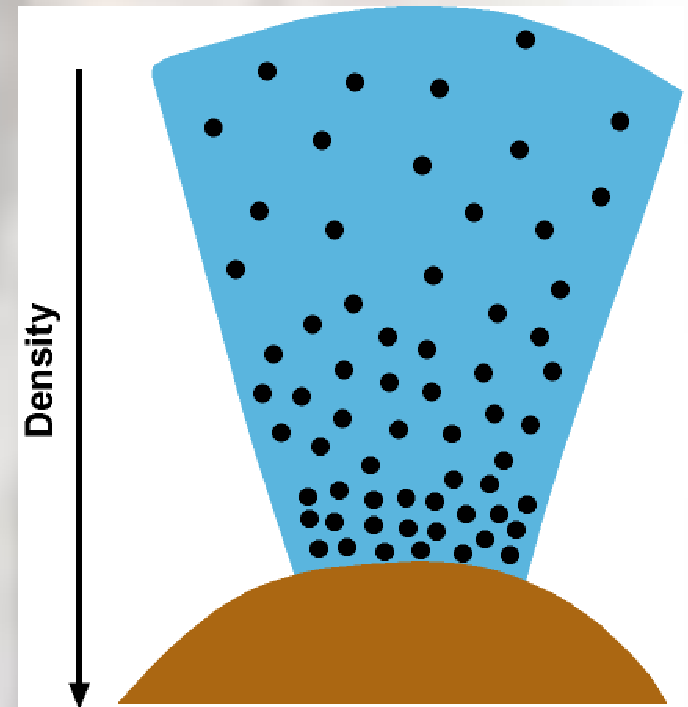


Density of Molecules

Gravity shapes and influences all atmospheric processes. It causes the **density** and **pressure** of air to decrease exponentially as one moves away from the surface of the Earth

About 97% of the atmosphere is within 30 km of the surface

75% is compressed below 10 km which is to say, within the troposphere



The pressure vs temperature vs density relates like this:

Ht. (M)/Press.-----Temp (in K)----- Density/kg m-3

0m, 1013.2mb-----288.2-----1.23

5km, 540.5mb-----255.7-----0.736

10km, 265.0mb-----223.2-----0.414

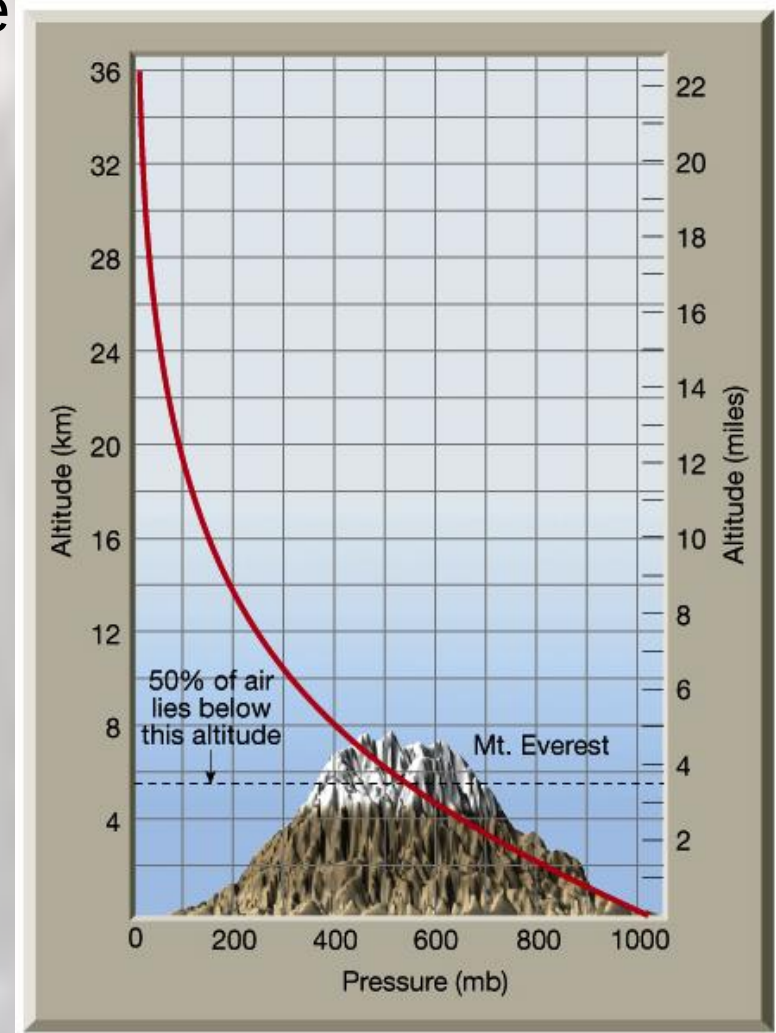
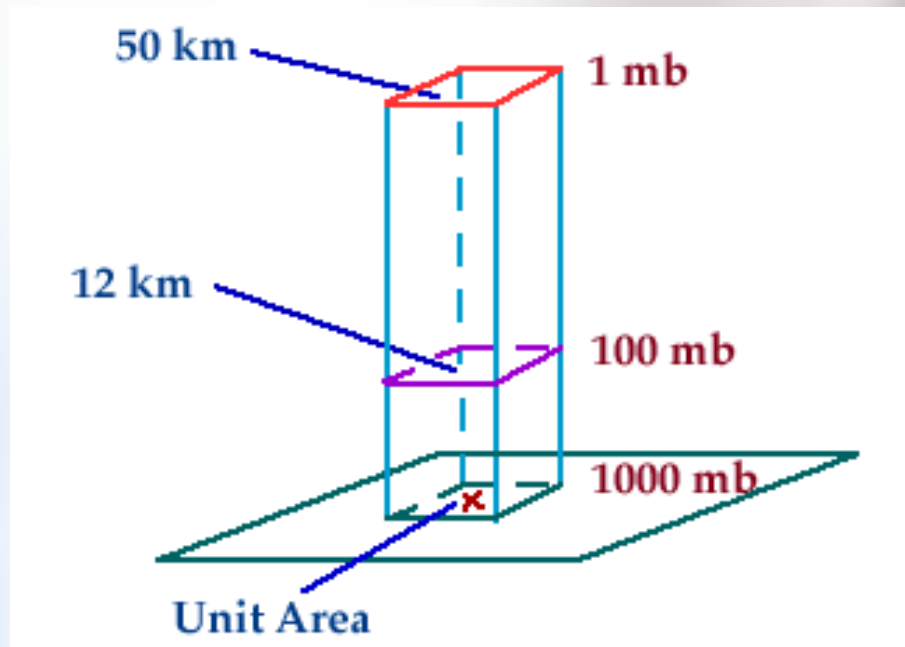
15km, 121.1mb-----216.7-----0.195

20km, 55.3mb-----216.7-----0.0889

Pressure with Height

pressure decreases with increasing altitude

The number of air molecules above a surface changes as the height of the surface above the ground changes.

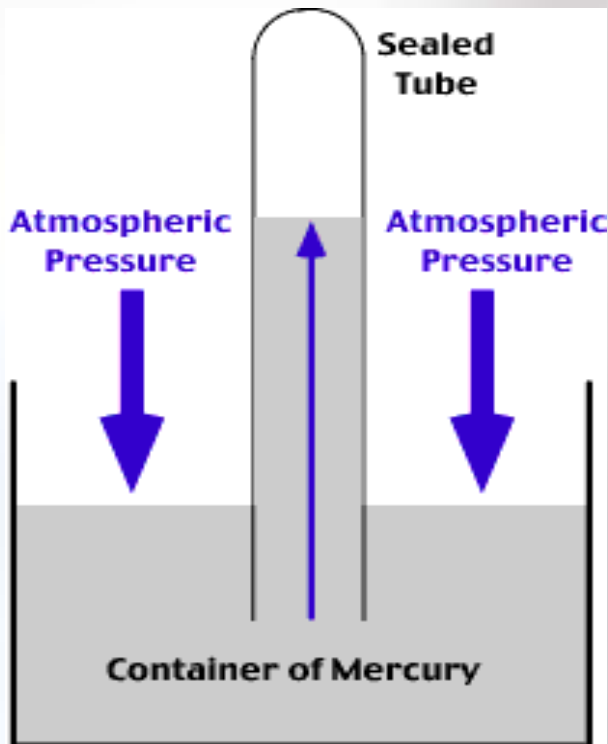


Pressure Measurement

Torricelli's Barometer

Weight of column of air above your head.

We can measure the density of the atmosphere by measuring the pressure it exerts.



**standard sea-level
pressure from
Torricelli's experiment
is 76.0 cm or 29.92
inches or 1013
millibars**